

$$\boxed{7-30} \quad |S| = \sqrt{s(s+1)}\hbar = \sqrt{3/4}\hbar \quad r = 10^{-15} \text{ m}, \quad m_e = 9.11 \cdot 10^{-31} \text{ kg}$$

PRETEND THAT SPIN, $|S|$, BEHAVES LIKE CLASSICAL (ORBITAL) ANGULAR MOMENTUM $|L| = I\omega = \left(\frac{2}{5}mr^2\right)\left(\frac{v}{r}\right)$

$$\Rightarrow v = |L|(5/2)/mr \quad |L| = \sqrt{3/4}\hbar = 9.13 \cdot 10^{-35} \text{ J}\cdot\text{s}$$

$$\Rightarrow \boxed{v = 2.5 \cdot 10^{11} \text{ m/s} \approx 1000c}$$


$$\boxed{7-32} \quad F_z = \mu_B \frac{dB}{dz} = -m_s g_L \mu_B \frac{dB}{dz} \quad F_z = m_{Ag} a_z \quad a_z \text{ is positive for } m_s = -1/2$$

$$\Rightarrow a_z = -\frac{m_s g_L \mu_B}{m_{Ag}} \frac{dB}{dz} \quad \text{or} \quad \frac{dB}{dz} = -\frac{a_z m_{Ag}}{m_s g_L \mu_B}$$

TO FIND a_z , WE LOOK AT THE KINEMATICS OF THE VERTICAL MOTION:

$$\text{TOTAL DEFLECTION} = d_1 + d_2 = .5 \text{ mm} = .5 \cdot 10^{-3} \text{ m}$$

$$d_1 = \frac{1}{2} a_z t^2$$

 FIND t FROM HORIZONTAL MOTION:

$$x_1 = v_x t \Rightarrow t = (1 \text{ m}) / 250 \text{ m/s} = 4 \cdot 10^{-3} \text{ s}$$

$$\Rightarrow d_1 = \frac{1}{2} a_z (4 \cdot 10^{-3} \text{ s})^2$$

$$d_2 = v_z t \quad \text{FIND } v_z = a_z t = (4 \cdot 10^{-3} \text{ s}) a_z \Rightarrow d_2 = a_z (4 \cdot 10^{-3} \text{ s})^2$$

$$\Rightarrow d_1 + d_2 = \left(\frac{3}{2}\right) a_z (4 \cdot 10^{-3} \text{ s})^2 = .5 \cdot 10^{-3} \text{ m} \quad \text{SOLVE FOR } a_z:$$

$$a_z = 20.8 \text{ m/s}^2 \quad M_{Ag} = 107.9 \frac{\text{kg}}{\text{mol}} \cdot \frac{\text{mol}}{6.022 \cdot 10^{23} \text{ atoms}} = 1.791 \cdot 10^{-25} \text{ kg}$$

$$\Rightarrow \frac{dB}{dz} = \frac{+(20.8 \text{ m/s}^2)(1.791 \cdot 10^{-25} \text{ kg})}{(+1/2)(2) \cdot 9.27 \cdot 10^{-24} \text{ J/T}} = \boxed{1.61 \text{ T/m}}$$

$\boxed{7-33}$ (a) $j = 3/2$ THERE ARE $2j+1$ DEFLECTIONS, SO THE # OF LINES IS $2(3/2)+1 = \boxed{4}$

(b) $l=1, s=0 \Rightarrow j=1$ SO # OF LINES = $2(1)+1 = \boxed{3}$

$\boxed{7-34}$ $l=1, 0 \quad s=1/2$ FOR $l=1, j=1/2, 3/2$
FOR $l=0, j=1/2 \Rightarrow 2^2S_{1/2}, 2^2P_{1/2}, 2^2P_{3/2}$

| | | |
|--|--|---|
| FOR $n=4, l = \left\{ \begin{array}{l} 3 \\ 2 \\ 1 \\ 0 \end{array} \right.$ | $\left. \begin{array}{l} j=5/2, 7/2 \\ j=3/2, 5/2 \\ j=3/2, 1/2 \\ j=1/2 \end{array} \right\}$ | $\left. \begin{array}{l} 4^2F_{5/2} \quad 4^2F_{7/2} \\ 4^2D_{3/2} \quad 4^2D_{5/2} \\ 4^2P_{1/2} \quad 4^2P_{3/2} \quad 4^2S_{1/2} \end{array} \right\}$ |
|--|--|---|

$$\boxed{7-37} \quad \text{if } j = l + 1/2 \text{ then } l = 1$$

$$\text{if } j = l - 1/2, \text{ then } l = 2$$

$$\boxed{7-40} \quad (a) \quad E = h\nu \quad \nu = c/\lambda \Rightarrow E = hc/\lambda$$

$$E_1 = \frac{(4.135 \cdot 10^{-15} \text{ eV} \cdot \text{s})(3 \cdot 10^8 \text{ m/s})}{5.89 \cdot 6 \cdot 10^{-9} \text{ m}} = 2.104 \text{ eV}$$

$$E_2 = \frac{(4.135 \cdot 10^{-15} \text{ eV} \cdot \text{s})(3 \cdot 10^8 \text{ m/s})}{589 \cdot 10^{-9} \text{ m}} = 2.106 \text{ eV}$$

$$(b) \quad \Delta E = .002 \text{ eV}$$

$$(c) \quad \mu_B = 5.79 \cdot 10^{-5} \text{ eV/Tesla} \quad \Delta E = 2\mu_B B \Rightarrow B = \frac{\Delta E}{2\mu_B}$$

$$= \frac{(.001 \text{ eV}) \text{ Tesla}}{(5.79 \cdot 10^{-5} \text{ eV})} = \boxed{17.3 \text{ T}}$$

$\boxed{7-43}$ (a) GROUND STATE: PUT 2 e^- IN $n=1$, 2 IN $n=2$, 1 IN $n=3$

(CANNOT PUT ALL IN $n=1$ STATE DUE TO PAULI EXCLUSION)

$$E = 2E_1 + 2E_2 + E_3 \quad E_n = \frac{\hbar^2 \pi^2 n^2}{2m_e L^2}$$

$$= \frac{\hbar^2 \pi^2}{2m_e L^2} [2(1)^2 + 2(2)^2 + 3^2] = \frac{19 \hbar^2 \pi^2}{2m_e L^2}$$

$$= \frac{19 (\hbar c)^2 \pi^2}{2(m_e c^2) L^2} \quad \hbar c = 197.3 \text{ eV} \cdot \text{m} \quad = \boxed{7.14 \text{ eV}}$$

$$m_e c^2 = .511 \cdot 10^6 \text{ eV}$$

(b) PIONS MUST BE BOSONS IF THEY HAVE SYMMETRIC WAVEFUNCTIONS, SO THEY ARE NOT SUBJECT TO PAULI EXCLUSION.

$$E = \frac{5E_1}{26A} = \boxed{.0071 \text{ eV}}$$

WHERE E_1 IS FOR AN ELECTRON.

$$\boxed{7-48} \quad E_n = -Z_{\text{eff}}^2 E_1 / n^2 \quad E_n = q_e V = (-e)(5.14 \text{ V})$$

$$(e)(\text{volts}) = 1 \text{ eV} \Rightarrow E_n = -5.14 \text{ eV} \stackrel{\text{set}}{=} Z_{\text{eff}}^2 (-1.51 \text{ eV})$$

$$\Rightarrow Z_{\text{eff}} = \sqrt{5.14 / 1.51} = 1.84$$

$$\boxed{7-49} \quad (a) \quad 1s^2 2s^2 2p^6 3s^2 3p^2$$

of electrons = $2 + 2 + 6 + 2 + 2 = 14$

ATOMIC # 14 \rightarrow SILICON

$$(b) \quad 1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 \quad \# \text{ of } e^- = 20$$

CALCIUM HAS ATOMIC # 20.

HW 1 ERRATA

I FORGOT TO INCLUDE PROBLEM 7.11 IN THE SOLUTIONS

$$\boxed{7.11} \quad (a) \quad \omega = 2\pi (735) \frac{\text{rev}}{\text{min}} \cdot \frac{\text{min}}{60 \text{ s}} = 77.0 \text{ rev/s}$$

$$L = I\omega = 10^{-5} \omega = \boxed{7.7 \cdot 10^{-4} \text{ kg m}^2 / \text{s}}$$

$$(b) \quad L = \sqrt{l(l+1)} \hbar \Rightarrow l(l+1) = \frac{L^2}{\hbar^2} = 5.322 \cdot 10^{61}$$

SINCE l IS LARGE, $l \approx l+1$ SO CAN WRITE

$$l(l+1) \approx l^2 = 5.322 \cdot 10^{61} \Rightarrow \boxed{l \approx 7.30 \cdot 10^{30}}$$